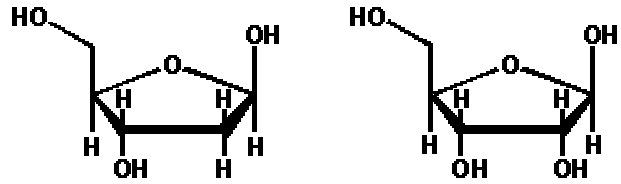


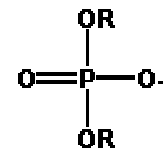
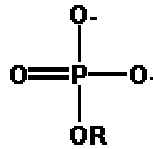
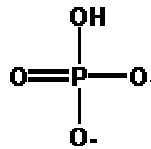


- Sugars.
  - DNA contains deoxyribose
  - RNA contains ribose



**D-2-Deoxyribose**

**D-Ribose**



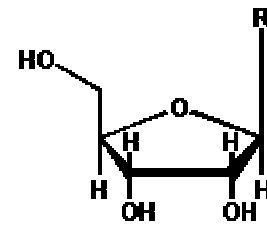
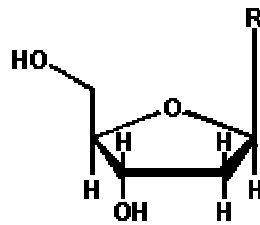
**Inorganic Phosphate**

**Phosphate monoester**

**Phosphate diester**

- Phosphate

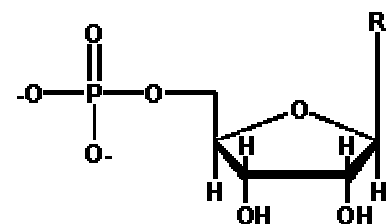
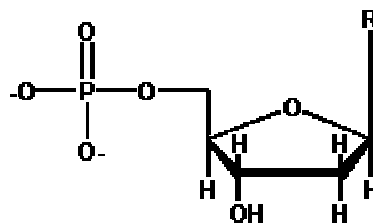
- In DNA, the bases are linked by an N-1'-glycosidic bond to **2'-deoxyribose** to form a **deoxynucleoside**.
- In RNA, the bases are linked by an N-1'-glycosidic bond to **ribose** to form a **nucleoside**.



**Deoxyribonucleoside**

**Ribonucleoside**

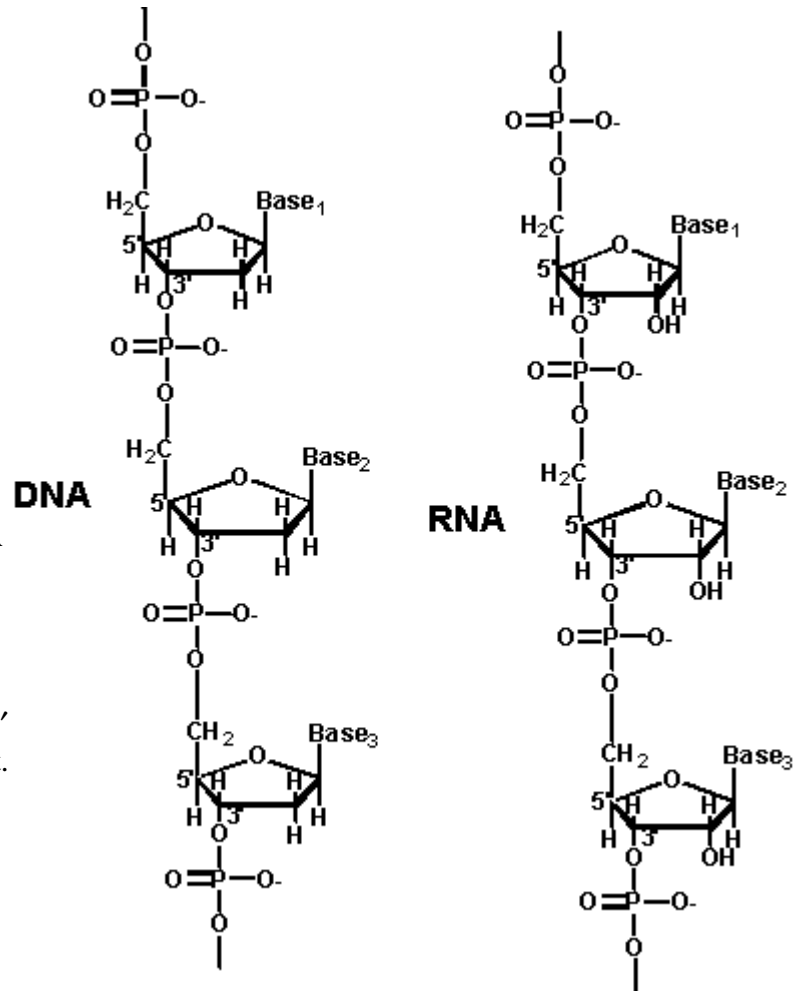
- **Deoxyribonucleotides** are **5' - monophosphate ester derivatives** of deoxynucleosides.
- **Ribonucleotides** are **5' - monophosphate ester derivatives** of nucleosides.



**5'-deoxyribonucleotide**

**5'-ribonucleotide**

**Nucleic Acids are Polymers of Nucleotide Phosphates**



- The polymers, DNA or RNA, are formed by **phosphodiester bonds** between the 5' hydroxyl of one nucleotide and the 3' hydroxyl of the next.

- The nucleotides are strong acids: the two  $pK_a$ s of the phosphomonoester are between 0.7 and 1.0 and between 6.1 and 6.3.
- Thus, in RNA and DNA the sugar phosphate backbone is negatively charged.
- The bases all absorb light strongly in the near ultraviolet ( $\sim 260$  nm).
- The phosphodiester bonds in nucleic acids are thermodynamically unstable ( $\Delta G$  for hydrolysis is about  $-25$  kJ/mol), but in the absence of a catalyst (enzyme) the polymers are quite stable.